

Towards rovibrationally resolved photochemistry in H₂O-CO₂

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Water and CO₂ are the most important greenhouse gases in our atmosphere. The solvation of CO₂ is a decisive process in the chemistry of clouds and oceans. A precise characterization of the interaction of these two molecules is thus of prime importance. In the present work, we recorded a rotationally resolved spectrum of the triple excitation of the OH stretch in the H₂O-CO₂ molecular complex. This represents a further step in our longstanding effort to better characterize the dynamics of CO₂-H₂O interactions, notably through a systematic increase in vibrational excitation [1–4]. Increasing excitation will ultimately drive the isomerization of the complex into carbonic acid H₂CO₃. The complexes were formed from an 8 cm-long pulsed slit supersonic jet and probed by the CRDS technique in the spectral range of the second OH overtone of H₂O, using the FANTASIO setup [5,6]. A home-built external cavity diode laser (ECDL), stabilized to the frequency comb, was used as the light source. The recorded spectrum was vibrationally assigned to the $(v_1, v_2, v_3) \leftarrow (v_1', v_2', v_3') = (2, 0, 1) \leftarrow (0, 0, 0)$ and $(2, 0, 1) + \text{intermolecular mode} \leftarrow (0, 0, 0)$ where the v_1, v_2, v_3 vibrational quantum numbers of the isolated H₂O molecule are used to assign the involved states. In this talk, I will present the experimental spectrum and the improvements to the experimental setup that enabled the recording of this spectral signature. I am going to present our progress in the analysis of this spectrum using group theory and the effective Hamiltonian. Finally, perspectives on going across the isomerization barrier will be presented.

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