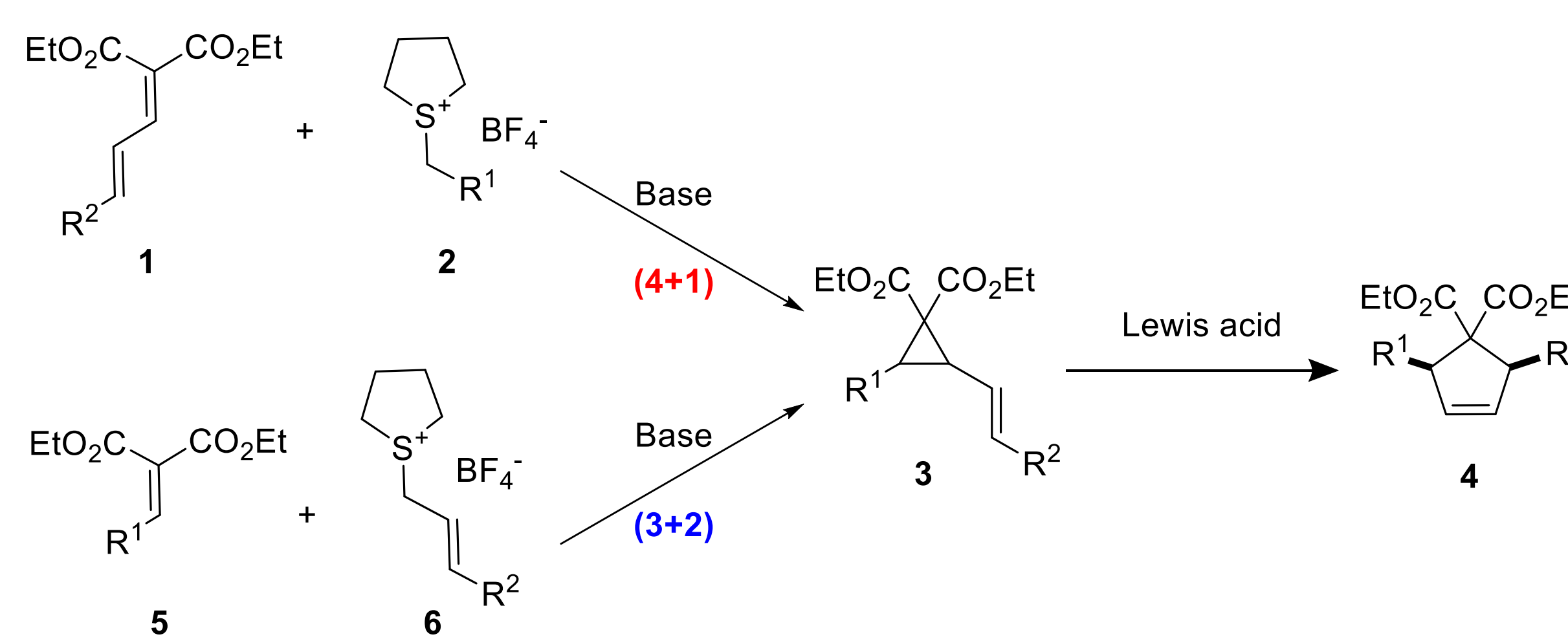


# New formal (4+3) annulation strategy by taking advantage of cyclopropanes reactivity

## - Background -

Our research group has been working these last years on the development of annulation strategies using sulfonium ylides. We have successfully developed effective methodologies for synthesizing five-membered carbocycles **4**, through formal (4+1) and (3+2) annulation strategies.<sup>[1,2]</sup> In both pathways, we initiate the process by forming a vinylcyclopropane intermediate **3**. This intermediate is then rearranged into **4**, facilitated by a Lewis acid catalysis.

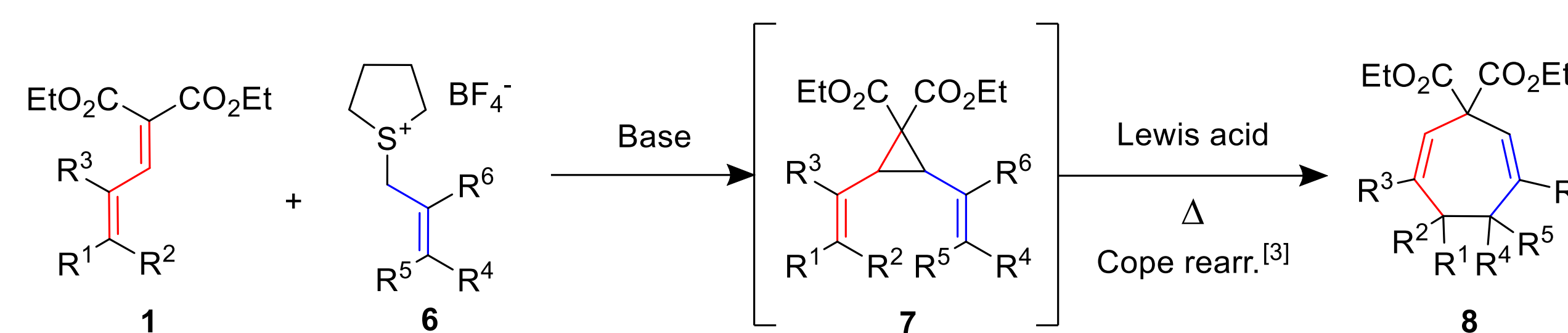


## - Introduction and objectives -

We would like to continue the exploitation of cyclopropane's reactivity to develop new annulation methodologies. The aim of this project is to obtain a series of seven-membered carbocycles **8** through a new formal (4+3) annulation strategy. To achieve this, we propose using activated olefins **1** from the (4+1) strategy and allylic sulfonium salts **6** from the (3+2) annulation.

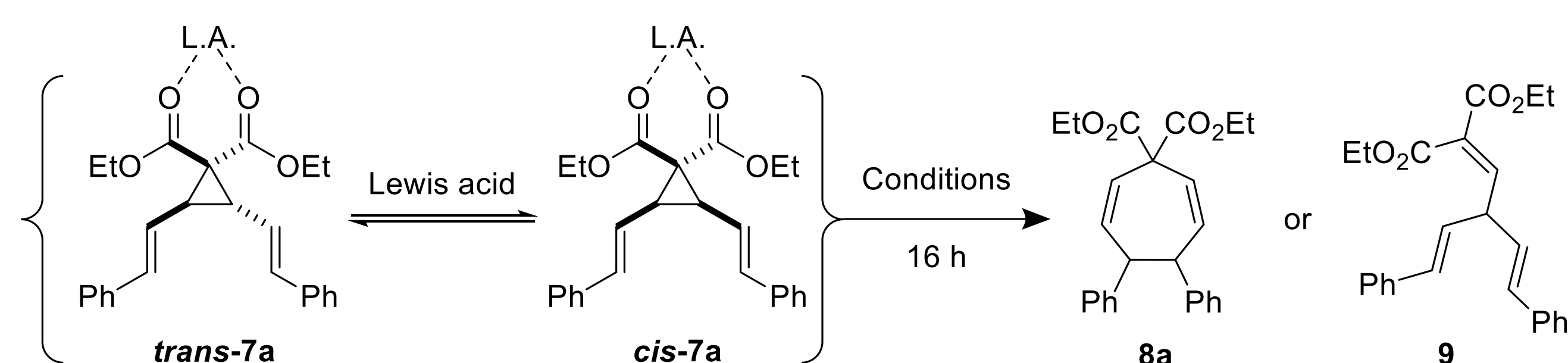
In practice, this work is leaded by two main objectives :

1. Develop and optimise the conditions of the Cope rearrangement (2<sup>nd</sup> step)
2. Determine the scope of this formal (4+3) strategy



## - Results (1) : Cope optimisation -

Addition of Lewis acid helps for the *cis*  $\rightleftharpoons$  *trans* isomerization  $\Rightarrow$  better conversion and decreases working T°



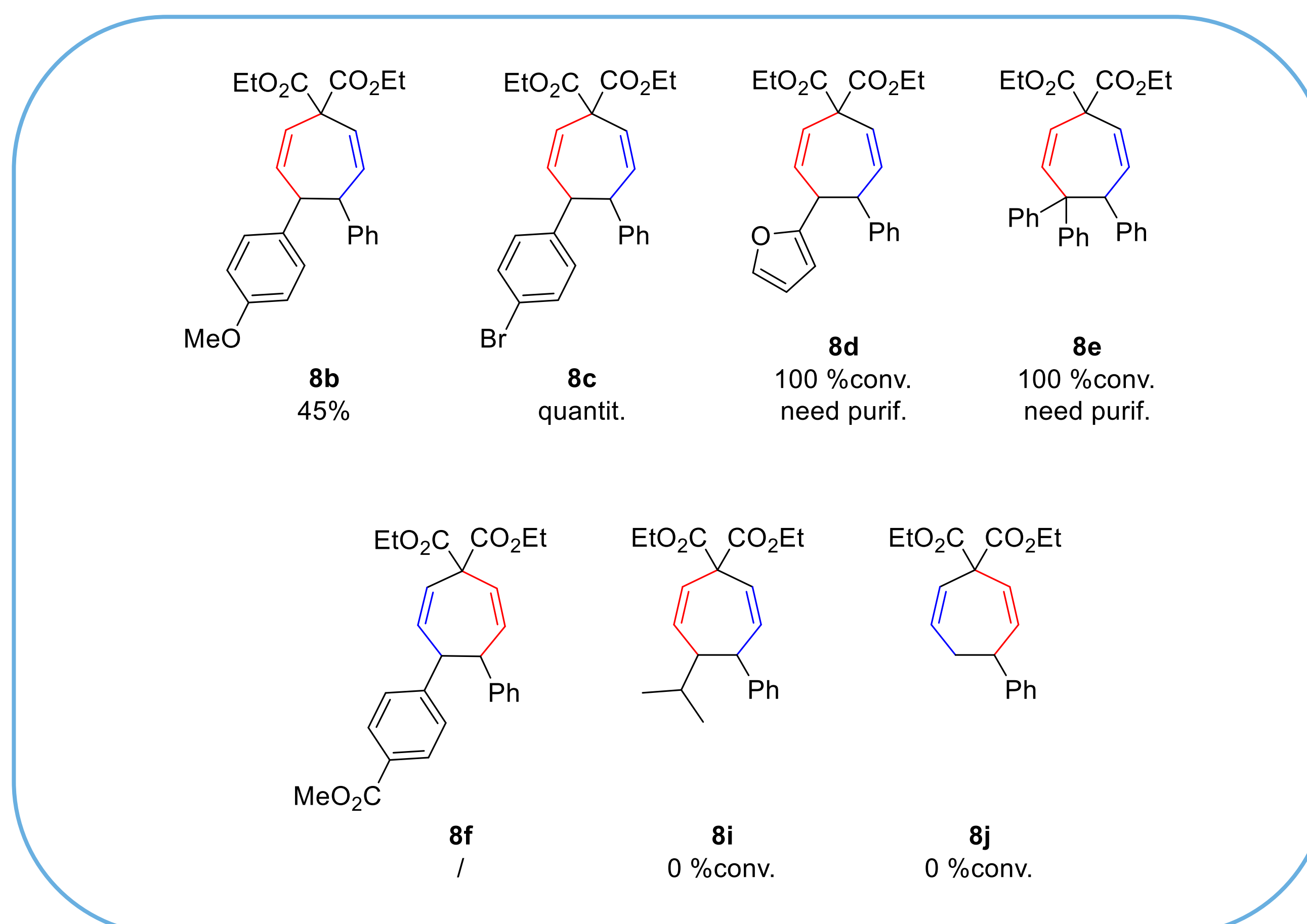
| Entry | Lewis ac. (eq.)            | T° (°C) | Solvent | trans-7a : cis-7a : 8a ratio |
|-------|----------------------------|---------|---------|------------------------------|
| 1     | /                          | 130     | Xylene  | 0 : 0 : 100                  |
| 2 [4] | /                          | 110     | Toluene | 30 : 0 : 70                  |
| 3 [4] | ZnCl <sub>2</sub> (0.1)    | 110     | Toluene | < 1 : < 1 : 99               |
| 4     | Fe(OTf) <sub>3</sub> (0.1) | 110     | Toluene | Degradation                  |
| 5     | Sc(OTf) <sub>3</sub> (0.1) | 110     | Toluene | Degradation                  |
| 6     | TiCl <sub>4</sub> (0.1)    | 110     | Toluene | <b>9</b> only                |
| 6     | MgI <sub>2</sub> (0.1)     | 80      | MeCN    | 30 : 0 : 70                  |
| 7*    | MgI <sub>2</sub> (0.3)     | 80      | MeCN    | < 5 : < 1 : > 95             |
| 8     | MgI <sub>2</sub> (0.1)     | r.t.    | MeCN    | 100 : 0 : 0                  |

\* 24 h instead of 16 h

Interestingly, the rearrangement was successfully performed with unpurified **7a**, which will increase the global yield of this strategy.

## - Results (2) : scope -

A series of cycloheptadienes with both electro-donating or electro-withdrawing aryl substituents (**8b** to **8d**) was obtained with a total conversion of the divinylcyclopropane (regardless of the initial *cis:trans* ratio).



- Compatible with heteroaromatic substituent (**8d**)
- Formation of a sterically hindered quaternary carbon (**8e**)
- Electron deficient aryl substituents prevent the cyclopropanation (**8f**)
- Alkyl substituents and terminal alkene prevent the rearrangement (**8i** and **8j**)

## - Conclusions -

We successfully develop an efficient new formal (4+3) annulation strategy using sulfonium ylides through the rearrangement of divinylcyclopropane derivative. We observed that a thermal activation is mandatory to perform the rearrangement of *cis*-divinylcyclopropane and the *cis*  $\rightleftharpoons$  *trans* isomerization. As the rearrangement takes place at lower temperature than the isomerization, we have discovered that it is possible to decrease the working temperature with the addition of a Lewis acid (MgI<sub>2</sub>) in a catalytic amount to obtain a total conversion.

In addition, a series of cycloheptadienes was already obtained with electro-donating or electro-withdrawing group, even heteroaromatic (2-furyl) but, at this time, we are not able to obtain the divinylcyclopropane intermediate with an electron-depleted aryl and the rearrangement seems to be prevented with alkyl substituents or terminal alkene (no substituent on one reagent). Other examples are planned, such as a second heteroaromatic group (2-pyridine) or *m*-MeO-C<sub>6</sub>H<sub>5</sub>.

## References:

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