

Mouhamadou S. Boye,<sup>a</sup> Aminata Diassé-Sarr,<sup>a\*</sup> Kieran C. Molloy,<sup>b</sup> Gabriele Kociok-Köhn<sup>b</sup> and Bernard Mahieu<sup>c</sup>

<sup>a</sup>Département de Chimie, Faculté des Sciences et Techniques, Université cheikh Anta Diop, Dakar, Senegal, <sup>b</sup>Département of Chemistry, University of Bath, Bath BA2 7AY, England, and <sup>c</sup>Département de Chimie, CSTR, Place Louis Pasteur B-1348, Louvain-la-Neuve, Belgium

Correspondence e-mail: diassem@yahoo.fr

#### Key indicators

Single-crystal X-ray study  
 $T = 150\text{ K}$   
 Mean  $\sigma(\text{C}-\text{C}) = 0.006\text{ \AA}$   
 $R$  factor = 0.031  
 $wR$  factor = 0.062  
 Data-to-parameter ratio = 29.3

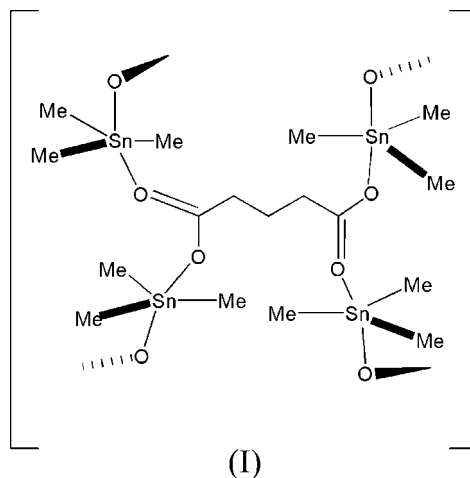
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

## Poly[bis(trimethyltin)- $\mu$ -glutarato]

The structure of  $[\text{SnMe}_3\text{O}_2\text{C}(\text{CH}_2)_3\text{CO}_2\text{SnMe}_3]_n$  or  $\{[\text{Sn}_2(\text{CH}_3)_6(\text{C}_5\text{H}_6\text{O}_4)]\}_n$  consists of an infinite layer containing pentacoordinate Sn atoms. There are two formula units, A and B, in the asymmetric unit. Glutarate acts as a tetradentate bridging ligand, each glutarate being linked to four different Sn atoms, generating two-dimensional sheets. One ligand links the Sn atoms of unit A, the other the Sn atoms of unit B. The sheets are parallel to each other.

#### Comment

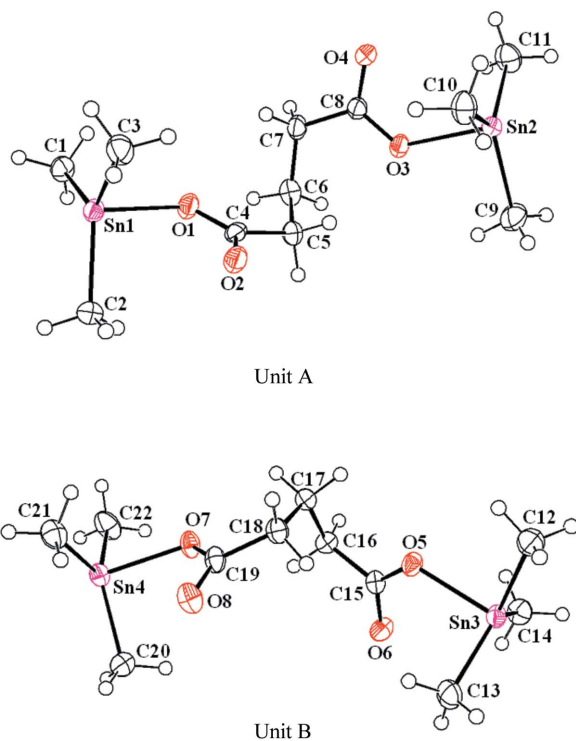
Crystal structures of  $[\text{SnR}_3]_2[\text{O}_2\text{C}(\text{CH}_2)\text{CO}_2]$  ( $R = \text{Me}$  and  $\text{Ph}$ ) and  $[\text{SnPh}_3]_2[\text{O}_2\text{C}(\text{CH}_2)_3\text{CO}_2]$  have been reported (Yin *et al.*, 2002; Cissé *et al.*, 2003). In addition to their interest as novel organotin species, bactericidal and fungicidal activity of triorganotin compounds are quite well known (Dutrecq *et al.*, 1992; Moens *et al.*, 1992). Such species are also screened for their antitumour activity and several of them are found to be active *in vitro* (Gielen, 2003, 2005). As a continuation of our interest in triaryltin derivatives (Diassé-Sarr *et al.*, 1997, 2004; Diop *et al.*, 2002), the X-ray crystallographic analysis of the title compound, (I) (Fig. 1), has been carried out and the results are presented here.



Each Sn atom is five-coordinate and adopts trigonal-bipyramidal *trans*- $\text{O}_2\text{SnC}_3$  geometry. There are two formula units in the asymmetric unit. The  $\text{O}-\text{Sn}-\text{O}$  bond angles are in the range  $171.32(10)$ – $173.34(10)^\circ$  and are close to the ideal value, while  $\text{C}-\text{Sn}-\text{C}$  angles within the equatorial plane of the Sn atoms cover the range  $117.35(17)$ – $123.25(18)^\circ$  and confirm the near-planarity of the  $\text{SnC}_3$  groups revealed by the IR data. Each Sn atom has short  $[\text{Sn1}-\text{O1} = 2.214(3)\text{ \AA}]$  and long  $[\text{Sn1}-\text{O2}^i = 2.346(3)\text{ \AA}]$ ; symmetry code as in Table 1]

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**Figure 1**

The structures of the two formula units in the asymmetric unit of (I), with ellipsoids drawn at the 50% probability level. H atoms have been omitted.

Sn—O bond lengths, which correspond to two C—O bond lengths. The longer C—O bond [C4—O1 = 1.276 (5) Å] corresponds to the stronger Sn—O bond and the shorter C—O bond [C4—O2 = 1.244 (5) Å] to the longer (weaker) Sn—O bond (Table 1). In (I), the Sn—O bonds are slightly longer than those reported for the triphenyltin analogue (Yin *et al.*, 2002). The glutarate acts as a tetradentate bridging ligand, each glutarate being linked to four different Sn atoms, generating two-dimensional sheets. A first ligand links Sn1 and Sn2, another one Sn3 and Sn4. The sheets are parallel to each other (Fig. 2). The network incorporates 24-atom rectangular rings with approximate dimensions  $9.9 \times 6.7$  Å measured between Sn atoms on opposite edges of the rectangle.

## Experimental

Me<sub>4</sub>NOH (10%), HCO<sub>2</sub>(CH<sub>2</sub>)<sub>3</sub>CO<sub>2</sub>H and SnMe<sub>3</sub>Cl were obtained from Aldrich and were used without further purification. An ethanol solution (20 ml) containing [Me<sub>4</sub>N]<sub>2</sub>[CO<sub>2</sub>(CH<sub>2</sub>)<sub>3</sub>CO<sub>2</sub>]<sub>2</sub>·5H<sub>2</sub>O (0.37 g, 1 mmol) [obtained from a solution of Me<sub>4</sub>NOH (20% in water) and HCO<sub>2</sub>(CH<sub>2</sub>)<sub>3</sub>CO<sub>2</sub>H in 1:2 ratio] and SnMe<sub>3</sub>Cl (0.20 g, 1 mmol) was stirred at room temperature for more than 2 h to give a solution from which colourless crystals were formed after slow solvent evaporation (yield 80%; m.p. 391 K).

### Crystal data

|                                                                                                   |                                   |
|---------------------------------------------------------------------------------------------------|-----------------------------------|
| [Sn <sub>2</sub> (CH <sub>3</sub> ) <sub>6</sub> (C <sub>5</sub> H <sub>6</sub> O <sub>4</sub> )] | $V = 3404.83$ (9) Å <sup>3</sup>  |
| $M_r = 457.68$                                                                                    | $Z = 8$                           |
| Orthorhombic, $P2_12_12_1$                                                                        | Mo $K\alpha$ radiation            |
| $a = 10.0053$ (1) Å                                                                               | $\mu = 2.94$ mm <sup>-1</sup>     |
| $b = 13.1391$ (2) Å                                                                               | $T = 150$ (2) K                   |
| $c = 25.9006$ (5) Å                                                                               | $0.25 \times 0.25 \times 0.15$ mm |

### Data collection

Nonius KappaCCD diffractometer  
Absorption correction: multi-scan  
(SORTAV; Blessing, 1995)  
 $T_{\min} = 0.502$ ,  $T_{\max} = 0.641$

42835 measured reflections  
9389 independent reflections  
8432 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.058$

### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.031$   
 $wR(F^2) = 0.062$   
 $S = 1.04$   
9389 reflections  
320 parameters  
H-atom parameters constrained

$\Delta\rho_{\text{max}} = 0.72$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -1.11$  e Å<sup>-3</sup>  
Absolute structure: Flack (1983),  
3866 Friedel pairs  
Flack parameter: 0.50 (2)

**Table 1**

Selected geometric parameters (Å, °).

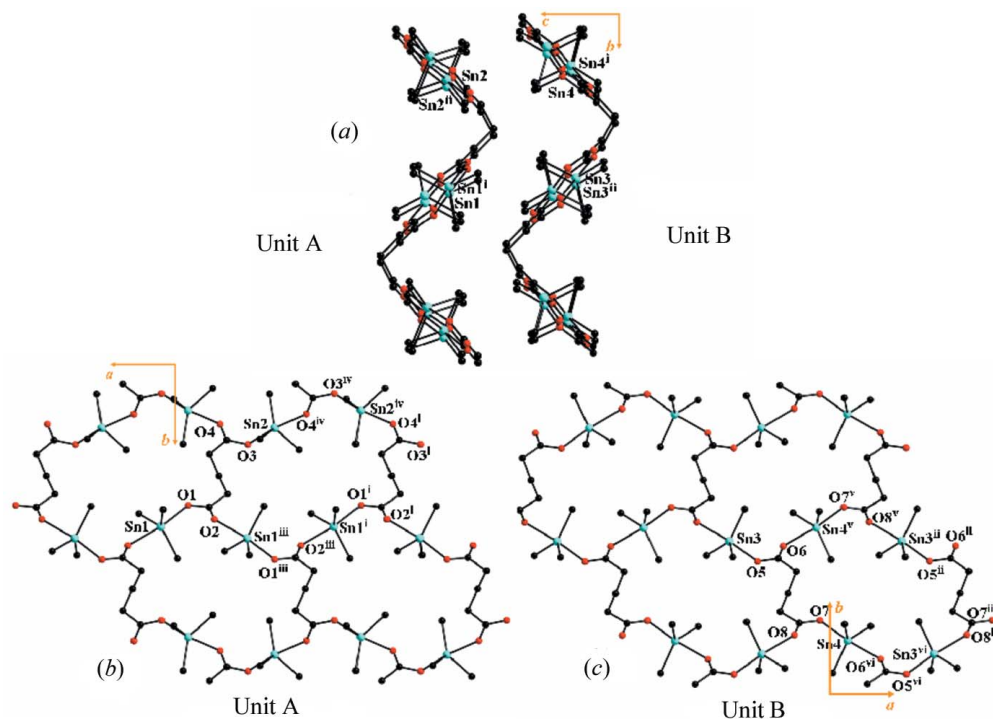
|                         |             |                           |             |
|-------------------------|-------------|---------------------------|-------------|
| Sn1—O1                  | 2.214 (3)   | O2—C4                     | 1.244 (5)   |
| Sn1—O2 <sup>i</sup>     | 2.346 (3)   | O3—C8                     | 1.270 (5)   |
| Sn2—O3                  | 2.201 (3)   | O4—C8                     | 1.258 (5)   |
| Sn2—O4 <sup>ii</sup>    | 2.333 (3)   | O5—C15                    | 1.276 (5)   |
| Sn3—O5                  | 2.197 (3)   | O6—C15                    | 1.243 (5)   |
| Sn3—O8 <sup>iii</sup>   | 2.343 (3)   | O7—C19                    | 1.268 (5)   |
| Sn4—O7                  | 2.211 (3)   | O8—C19                    | 1.255 (5)   |
| O1—C4                   | 1.276 (5)   |                           |             |
| C1—Sn1—C2               | 117.35 (17) | C10—Sn2—O3                | 94.60 (14)  |
| C1—Sn1—C3               | 119.91 (18) | C10—Sn2—O4 <sup>ii</sup>  | 88.79 (14)  |
| C3—Sn1—C2               | 122.34 (18) | C11—Sn2—O3                | 95.74 (15)  |
| C10—Sn2—C9              | 118.3 (2)   | C11—Sn2—O4 <sup>ii</sup>  | 88.60 (14)  |
| C11—Sn2—C10             | 122.02 (19) | C12—Sn3—O5                | 88.33 (14)  |
| C11—Sn2—C9              | 119.1 (2)   | C12—Sn3—O8 <sup>iii</sup> | 85.05 (14)  |
| C13—Sn3—C12             | 119.36 (19) | C13—Sn3—O5                | 94.11 (14)  |
| C14—Sn3—C13             | 123.19 (18) | C13—Sn3—O8 <sup>iii</sup> | 88.54 (14)  |
| C14—Sn3—C12             | 117.01 (19) | C14—Sn3—O5                | 93.89 (14)  |
| C20—Sn4—C21             | 123.25 (18) | C14—Sn3—O8 <sup>iii</sup> | 89.75 (14)  |
| C22—Sn4—C20             | 118.79 (19) | C20—Sn4—O6 <sup>iv</sup>  | 87.64 (14)  |
| C22—Sn4—C21             | 117.4 (2)   | C21—Sn4—O6 <sup>iv</sup>  | 89.90 (14)  |
| C1—Sn1—O1               | 86.64 (13)  | C20—Sn4—O7                | 95.28 (14)  |
| C1—Sn1—O2 <sup>i</sup>  | 85.01 (13)  | C21—Sn4—O7                | 94.41 (14)  |
| C2—Sn1—O1               | 95.15 (14)  | C22—Sn4—O6 <sup>iv</sup>  | 84.66 (14)  |
| C2—Sn1—O2 <sup>i</sup>  | 90.80 (14)  | C22—Sn4—O7                | 87.70 (14)  |
| C3—Sn1—O1               | 94.33 (14)  | O1—Sn1—O2 <sup>i</sup>    | 171.32 (10) |
| C3—Sn1—O2 <sup>i</sup>  | 87.79 (15)  | O3—Sn2—O4 <sup>ii</sup>   | 171.96 (10) |
| C9—Sn2—O3               | 87.29 (15)  | O5—Sn3—O8 <sup>iii</sup>  | 173.34 (10) |
| C9—Sn2—O4 <sup>ii</sup> | 84.68 (15)  | O7—Sn4—O6 <sup>iv</sup>   | 172.31 (10) |

Symmetry codes: (i)  $x + \frac{1}{2}, -y + \frac{3}{2}, -z + 2$ ; (ii)  $x - \frac{1}{2}, -y + \frac{1}{2}, -z + 2$ ; (iii)  $-x - 1, y + \frac{1}{2}, -z + \frac{3}{2}$ ; (iv)  $-x, y - \frac{1}{2}, -z + \frac{3}{2}$ .

The structure was refined as an inversion twin, with components in the ratio 1:1. All H atoms were geometrically located in ideal positions and refined using a riding model, with C—H = 0.98 Å for methyl H atoms and C—H = 0.99 Å for methylene H atoms, and with  $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$  for methyl H atoms and  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  for methylene H atoms. The deepest hole in the final difference map is 0.78 Å from Sn4.

Data collection: COLLECT (Nonius, 1997); cell refinement: HKL SCALEPACK (Otwinowski & Minor, 1997); data reduction: DENZO (Otwinowski & Minor, 1997); program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1998); molecular graphics: ORTEP-3 (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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**Figure 2**

(a) A view along the *a* axis of the structure of (I), showing the two units. (b) A view along the *b* axis of unit A, showing the two-dimensional sheet incorporating Sn1 and Sn2. (c) A view along the *b* axis of unit B, showing the two-dimensional sheet incorporating Sn3 and Sn4. [Symmetry codes: (i)  $x - 1, y, z$ ; (ii)  $x + 1, y, z$ ; (iii)  $x - \frac{1}{2}, \frac{3}{2} - y, 2 - z$ ; (iv)  $x - \frac{1}{2}, \frac{1}{2} - y, 2 - z$ ; (v)  $-x, \frac{1}{2} + y, \frac{3}{2} - z$ ; (vi)  $-x, y - \frac{1}{2}, \frac{3}{2} - z$ .]

## References

- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
- Blessing, R. H. (1995). *Acta Cryst.* **A51**, 33–38.
- Cissé, I., Guèye, O., Mahieu, B. & Tinant, B. (2003). *Appl. Organomet. Chem.* **17**, 821–822.
- Diassé-Sarr, A., Barry, A. H., Jouini, T., Diop, L., Mahieu, B., Mahon, M. F. & Molloy, K. C. (2004). *J. Organomet. Chem.* **689**, 2087–2091.
- Diassé-Sarr, A., Diop, L., Mahon, M. F. & Molloy, K. C. (1997). *Main Group Met. Chem.* **20**, 223–229.
- Diop, C. A. K., Bassène, S., Sidibé, M., Diassé-Sarr, A., Diop, L., Molloy, K. C., Mahon, M. F. & Toscano, R. A. (2002). *Main Group Met. Chem.* **25**, 683–689.
- Dutrecq, A., Willem, R., Biesemans, M., Boualam, M., Meriem, A. & Gielen, M. (1992). *Main Group Met. Chem.* **15**, 285–291.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Farrugia, L. J. (1999). *J. Appl. Cryst.* **32**, 837–838.
- Flack, H. D. (1983). *Acta Cryst.* **A39**, 876–881.
- Gielen, M. (2003). *J. Braz. Chem. Soc.* **14**, 870–877.
- Gielen, M. (2005). *Appl. Organomet. Chem.* **19**, 440–450.
- Moens, L., Maraite, H., Mahieu, B., El Khouloufi, A., Willem, R. & Gielen, M. (1992). *Main Group Met. Chem.* **15**, 275–291.
- Nonius (1997). *COLLECT*. Nonius BV, Delft, The Netherlands.
- Otwinowski, Z. & Minor, W. (1997). *Methods in Enzymology*, Vol. 276, *Macromolecular Crystallography*, Part A, edited by C. W. Carter Jr & R. M. Sweet, pp. 307–326. New York: Academic Press.
- Sheldrick, G. M. (1998). *SHELXL97*. University of Göttingen, Germany.
- Yin, H.-D., Wang, C.-H., Ma, C.-L. & Fang, H.-X. (2002). *Chin. J. Org. Chem.* **22**, 489–495.